PATENT COOPERATION TREATY

PCT

TRANSLATION INTERNATIONAL PRELIMINARY REPORT ON PATENTABILITY

(Chapter II of the Patent Cooperation Treaty)

(PCT Article 36 and Rule 70)

Applicant's or agent's file reference 4465-X-22934			ce	FOR FURTHER ACTION		See Form PCT/IPEA/416			
International application No.				International filing da	te (day/month/year)	Priority date (day/month/year)			
PCT/EP2004/012911			911	14.11.200	4	14.11.2003			
Internati	ional Pa	tent Classification	n (IPC) or natio	onal classification and	IPC				
C01	C01 B25/45, H01 M4/58, H01 M4/02								
	Applicant SÜD-CHEMIE AG								
1.						International Preliminary Examining Authority			
				e applicant according t 7					
2.		EPORT consists	_		sheets, including	g this cover sheet.			
3.	This re	eport is also acco	mpanied by Al	NNEXES, comprising:					
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		sheets				nmended and are the basis for this report and/or ale 70.16 and Section 607 of the Administrative			
			· · · · · · · · · · · · · · · · · · ·	ede earlier sheets, but	which this Authority con-	siders contain an amendment that goes beyond			
		the di Box.	sclosure in the	international applicat	ion as filed, as indicated	in item 4 of Box No. I and the Supplemental			
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				rative Instructions).	s marcated in the Supple	inclinal Box Relating to Sequence Listing (see			
4.	This re	eport contains inc	lications relation	ng to the following iter	ns:				
	\boxtimes	Box No. I	Basis of the	report					
		Box No. II	Priority						
		Box No. III	Non-establis	shment of opinion with	regard to novelty, invent	ive step and industrial applicability			
		Box No. IV	Lack of unit	y of invention					
	\boxtimes	Box No. V		atement under Article : l explanations supporti		lty, inventive step or industrial applicability;			
		Box No. VI	Certain docu	aments cited					
		Box No. VII	Certain defe	cts in the international	application				
		Box No. VIII	Certain obse	ervations on the interna	itional application				
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Facsimile No.					Telephone No.				

International application No.
PCT/EP2004/012911

Box	k No. I	Basis of the report	
1.		n regard to the language, this report is based on the international application in the language in which it was filed, unless otherwise ated under this item.	e
		This report is based on translations from the original language into the following language which is the language of a translation furnished for the purposes of: international search (Rule 12.3 and 23.1(b)) publication of the international application (Rule 12.4) international application (Rule 55.3 and (ap. 55.3))	_ ,
2.	rece	international preliminary examination (Rule 55.2 and/or 55.3) regard to the elements of the international application, this report is based on (replacement sheets which have been furnished to iving Office in response to an invitation under Article 14 are referred to in this report as "originally filed" and are not annexample. the international application as originally filed/furnished the description:	
		pages 1-32 as originally filed/furnish pages* received by this Authority on	
		pages* received by this Authority on	
	\square		_
		the claims: nos. 1-40 as originally filed/furnis	bad
		nos.* as amended (together with any statement) under Article nos.* received by this Authority on	
		nos.* received by this Authority on	
	\square		_
		the drawings:	
		sheets 1/3-3/3 as originally filed/furnish	aed
		sheets* received by this Authority on	_
		sheets* received by this Authority on	_
	닏	a sequence listing and/or any related table(s) – see Supplemental Box Relating to Sequence Listing.	
3.	Ш	The amendments have resulted in the cancellation of:	
		the description, pages	_
		the claims, nos.	_
		the drawings, sheets/figs	_
		the sequence listing (specify):	_
		any table(s) related to sequence listing (specify):	_
4.		This report has been established as if (some of) the amendments annexed to this report and listed below had not been made, they have been considered to go beyond the disclosure as filed, as indicated in the Supplemental Box (Rule 70.2(c)).	since
		the description, pages	_
		the claims, nos.	_
		the drawings, sheets/figs	_
		the sequence listing (specify):	_
		any table(s) related to sequence listing (specify):	
*	If ite	m 4 applies, some or all of those sheets may be marked "superseded."	

YES
_ NO
YES
_ NO
YES
_ NO

- 2. Citations and explanations (Rule 70.7)
 - 1. This report makes reference to the following documents:
 - D1: WO 02/083555 A (ZENTRUM FUER

 SONNENERGIE- UND WASSERSTOFF
 FORSCHUNG BADEN-WUERTTEMBE) 24 October

 2002 (2002-10-24)
 - D2: ARNOLD G ET AL: "Fine-particle lithium iron phosphate LiFePO4 synthesized by a new low-cost aqueous precipitation technique" JOURNAL OF POWER SOURCES, ELSEVIER SEQUOIA S.A. LAUSANNE, CH, Vol. 119-121, 1 June 2003 (2003-06-01), pages 247-251, XP004430175 ISSN: 0378-7753
 - D3: WO 02/099913 A (N.V. UMICORE S.A; WURM, CALIN; MORCRETTE, MATHIEU; GWIZDALA, SYLVAIN;) 12 December 2002 (2002-12-12)

2. Novelty

D1 is regarded as the prior art closest to the subject matter of claim 1. It discloses a

Box No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

method for producing binary, ternary, and quaternary lithium phosphates of the formula $\text{Li}(\text{Fe}_x\text{M}^1_y\text{M}^2_z)\,\text{PO}_4$. Precursor compounds of the elements Li, Fe, M¹ and/or M² are precipitated from an aqueous solution and the precipitate is dried in a non-oxidizing atmosphere and then tempered. The particle size analysis of the lithium phosphate obtained results in a very narrow particle size distribution with an average particle size (D₅₀) of less than 3 µm.

Therefore, the subject matter of claim 1 differs from the known method in that the precursor suspension is not dried and tempered immediately after the precipitate is obtained, but rather first subjected to a dispersion or grinding treatment.

Consequently, the subject matter of the method claims 1-26 and 36-40 is novel (PCT Article 33(2)).

Claim 27 relates to a product obtainable according to one of the method claims 1-26. The claim does not, however, contain any product features that would enable a person skilled in the art to differentiate the LiMPO₄ in the claim from the lithium metal phosphates in the prior art.

The description indicates that such a product

Box No. V Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

has a D_{90} particle value of at most 25 μm , a D_{50} value of less than 0.8 μm , and a D_{10} value of less than 0.35 μm (page 15, first paragraph).

D2 discloses a method for producing phase-pure, homogeneous, crystalline LiFePO₄. A precursor material is precipitated from an aqueous solution and then heated to a temperature of $650-800^{\circ}$ C. The product has the following particle size distribution: a D₉₀ particle value of approximately 15 µm, a D₅₀ value of approximately 7 µm, and a D₁₀ value of approximately 3 µm (figure 4).

D3 discloses a lithium-transition metal phosphate with an average particle size of less than 1 μm and a specific surface area of 2.84 m^2/g (table 1).

Therefore, claims 27-31 are novel.

Since the LiFePO $_4$ in claim 27 is novel, a composition containing it, its use as electrode material, and a secondary battery containing the composition are likewise novel.

Consequently, claims 32-35 are novel.

Box No. V

Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

3. Inventive step

The problem to be solved by the present application is that of providing a method for producing lithium metal phosphate that results in material suitable for the electrodes of rechargeable batteries.

The solution to this problem as proposed in the present application involves an inventive step for the following reasons (PCT Article 33(3)):

The main difference between the production method in the prior art and the method in the present application is that the precursor mixture and/or precursor suspension is subjected to a dispersion or grinding treatment. The prior art contains nothing that suggests such a treatment as relates to the production of lithium metal phosphate.

The examples indicate that the compound (LiFePO₄) produced according to the application shows better electrochemical properties, in particular at high charge and discharge rates, than a material produced according to the prior art.

The production method, the LiFePO₄ produced according to the application, its use as electrode material, and a secondary battery

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Box No. V	Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement
	containing a LiFePO4 composition produced
	according to the application all involve an
	inventive step. $Claims 1-40$ are regarded as
	inventive.